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(54) Title: LOW PH ANTIMICROBIAL FOOD COMPOSITION (57) Abstract A nutritionally balanced water soluble powdered food composition which, when mixed with water, has a low pH, extended shelf life, high antimicrobial activity, and which includes protein alpha-amino acids in solution or in suspension. The food composition utilizes a binary stabilizer system.		

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LOW PH ANTIMICROBIAL FOOD COMPOSITION

10 This invention relates to nutritionally balanced food compositions for ingestion along the digestive tract of a patient.

More particularly, the invention relates to nutritionally balanced liquid food compositions which have a low pH, extended shelf life, high antimicrobial activity, and which include protein in solution or in suspension.

15 In a further respect, the invention relates to a liquid food composition including a low pH protein stabilizer system which, when the food composition is heated to a high temperature to be sterilized, maintains its homogeneity.

20 In another respect, the invention relates to a liquid food composition which includes a low pH protein stabilizer system and exhibits unusually low aerobic and anaerobic bacterial activity for long periods of time at room temperature.

25 Liquid nutritionally balanced food compositions are known in the art. See, for example, my U. S. Patent No. 4,931,300 for "ANTIMICROBIAL FOOD COMPOSITION".

30 Liquid nutritionally balanced powdered food compositions like those described in my U. S. Patent No. 4,931,300 have several potential disadvantages. Protein tends to precipitate from liquid solutions which, like the food composition in U. S. Patent No. 4,931,300, have acidic pH values in the range of 2.0 to 5.5. In particular, protein
35 tends to precipitate from such liquid solutions when the solutions are heated to a high temperature to sterilize the solutions. Solutions with low pH values in the range of 2.0 to 5.5 are, however, often preferred because the acidity of the solutions normally provides a high level of

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antimicrobial activity. Food compositions like the compositions disclosed in U. S. Patent No. 4,931,300 are an exception and do not provide a high degree of antimicrobial activity. This is evidenced by the fact that the food composition in my U. S. Patent No. 4,931,300 must be refrigerated after it is reconstituted and must then be utilized within about seventy-two hours. Even though the seventy-two hour shelf life of the reconstituted food composition is relatively short, it is still substantially longer than the shelf life of other comparable food compositions. See, for example, U. S. Patent No. 4,112,123 to Roberts, where the shelf life of a comparable reconstituted refrigerated food composition is only about twenty-four hours. Another problem associated with acidic aqueous food compositions of the type described in U. S. Patent Nos. 4,112,123 and 4,931,300 is that identifying an appropriate stabilizer for the food composition is difficult. The stabilizer must be able to act quickly when the food composition is reconstituted as a drink, must not produce a composition which has excessive viscosity, must have an extended shelf life, and must be able to resist degradation due to the acidic nature of the food composition.

Accordingly, it would be highly desirable to provide a liquid food composition which would produce a low viscosity solution which has a pH in the range of about 2.0 to 6.5, has a high antimicrobial activity, has an extended shelf life at room temperature, and which prevents protein from precipitating or settling from solution when the solution is sterilized at high temperature.

Therefore, it is a principal object of the invention to provide an improved food composition.

Another object of the invention is to provide a low pH liquid food composition which includes alpha-amino acids or other protein and which generally prevents protein from precipitating or separating from the liquid food composition.

A further object of the invention is to provide a

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nutritionally balanced liquid food composition which includes a low pH protein stabilizer system which has a high antimicrobial activity and has an extended shelf life at room temperature.

5 These and other, further and more specific objects and advantages of the invention will be apparent to those skilled in the art from the following detailed description thereof.

10 Briefly, I have discovered a food powder composition which has a high antimicrobial activity and extended shelf life. The food composition includes from 6% to 28% by weight of a water soluble protein; from 4 to 22% by weight of triglycerides of predominantly 6 to 26 carbon atoms in the fatty acid chain; from 35% to 78% by weight of
15 carbohydrates selected from the group consisting of corn syrup solids, trisaccharides, tetrasaccharides, pentasaccharides, hexasaccharides, dextrose, fructose, sucrose, maltose, oligosaccharides and high saccharides; from 0.01% to 10.0% by weight of an emulsifier; from 0.1% to
20 8% by weight of an edible acid; from 0.01% to 6% by weight of an antimicrobial agent selected from the group consisting of sorbic acid, benzoic acid, sodium benzoate, potassium sorbate, sodium sorbate, and potassium benzoate; and, from 0.1% to 20.0%, preferably 0.2% to 5.0% by weight of pectin.
25 The food composition provides up to about three calories per cubic centimeter of composition. On being reconstituted with water, the composition forms a liquid solution which has an osmolarity of 250 to 650. The pH of the reconstituted food composition is from 2.0 to 6.5, preferably 3.0 to about 5.7.
30 The water soluble protein is preferably whey protein or any other acid stable protein or peptide. The water soluble protein forms a sol in water. A sol is a fluid colloidal system, i.e., a dispersion of solid particles in a liquid colloidal solution.

35 The water soluble protein can be a whole protein or can be a partially hydrolyzed protein such as a protein alpha-amino acid. For purposes of the present specification, the term protein alpha-amino acids is defined

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to include any one or more of (a) mono peptides, dipeptides, tripeptides, and/or oligopeptides prepared by the partial hydrolysis of naturally occurring proteins or of artificially produced proteins, (b) mono peptides, dipeptides, tripeptides, and oligopeptides prepared by synthesis and comparable to such peptides prepared by the partial hydrolysis of natural or artificially produced proteins, and (c) whey protein. Whey protein and other naturally occurring proteins are presently preferred in the practice of the invention because of their ready availability and because they typically produce drinks which taste good. Protein alpha-amino acids do not include amino acids which are not linked or bonded to at least one other amino acid. In addition to or in place of protein, amino acids can be utilized in the food composition of the invention which are separate from one another and are not bonded or linked together to form protein.

Peptide alpha-amino acids help in reducing the pH of the food composition, and consequently, in reducing the quantity of acidulant required in preparing the food composition. A peptide is any of a class of amides that are derived from two or more amino acids by combination of the amino group of one acid with carboxyl group of another, that yield these acids on hydrolysis, that are classified according to the number of component amino acids, and that are obtained by partial hydrolysis of naturally occurring or artificially produced proteins or by synthesis (as from alpha-amino acids or their derivatives). A dipeptide is a peptide that yields two molecules of amino acid on hydrolysis. A polypeptide is a polyamide that yields amino acids on hydrolysis but has a lower molecular weight than a protein and that is obtained by partial hydrolysis of proteins or by synthesis. Peptides are easier to digest than whey and other proteins.

Peptides are prepared from hydrolyzing proteins of any kind, and are commonly prepared by hydrolyzing egg, milk, or soy. The proteins in egg, milk, and soy are examples of naturally occurring proteins.

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For purposes of the present specification, the term "whey protein" is defined to mean that water soluble or suspendible, essentially undenatured protein fraction derived from cheese whey which protein fraction is, essentially, retained by an ultra-filtration membrane that permits lactose, lactic acid, and soluble salts to pass through the membrane. Whey protein is a naturally occurring protein and is specific and identifiable in terms of its composition and is not necessarily dependant upon a process for production thereof. Whey protein may be obtained by methods other than ultra-filtration, e.g., gel filtration.

The amount of protein alpha-amino acids or other water soluble proteins used in the present powder food composition may vary widely, but for most applications from 4% to 22% on a dry weight basis is suitable, especially between about 15% and 20%.

The protein alpha-amino acids are essentially water soluble or suspendible, and capable of being compounded for formulated into stable and pourable form in order to function in the manner required. Further, it is the protein alpha-amino fraction containing one or more of the twenty alpha-amino acids, most of which have the general formula $RCH(NH_2)COOH$, that are synthesized in plant and animal tissues, that are considered the building blocks of proteins, from which they can be obtained by hydrolysis, and that play an important role in metabolism, growth, maintenance and repair of tissue.

Table 1 in U. S. Patent No. 4,112,123 to Roberts shows a typical amino acid profile for whey protein used in the present invention.

The alpha-amino acid proteins or whole protein utilized in the practice of the invention are in the form of amino acids each of which is bonded to one or more other amino acids. Such alpha-amino acids are typically natural proteins, but can be "directly" produced by synthesis. As used herein, natural protein is whole protein found in a plant or is protein produced by hydrolyzing one a whole protein which is found in a plant.

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Medium-chain triglycerides (MCT's) utilized in the food composition of the invention produce a composition of low viscosity while concomitantly providing high caloric content and easily digestible compositions. The fatty acid chains of medium-chain triglycerides are predominantly between about 6 and 12 carbon atoms, and are preferably utilized in conjunction with long-chain triglycerides (LCT's) in which fatty acid chains are predominately between about 14 to 26 carbon atoms.

The proportion of LCT's and MCT's in the powder food composition can vary widely, but typically is about 4% to 22% by weight, with 12% to 18% being an optimal range.

Any food grade emulsifier is used for present emulsification purposes and combinations for emulsifiers are used if desired. Any of the long fatty acid glycerol emulsifiers can be used, which normally have a C-12 to C-20 esterified chain. Typical among these are glycerol-lactopalmitate or the stearate. Alternately, the propylene derived emulsifiers may be used, such as propylene glycomonostearate, or the oleate, palmitate, and myristate. Likewise, the "Span" series of emulsifiers may be used. These are well-known emulsifiers and are fatty acid partial esters of the sorbitol anhydrides (or sorbitan). One well known emulsifier is the "Tween" series of polyoxyethylene derivatives of fatty acid partial esters of sorbitol anhydride. Tween 80 and Atmos 300 are often used in combination. The well known Atmos series of mono and diglycerides may be used. Also, lecithin is a suitable emulsifier. The amount of the emulsifier is chosen to suit the particular powder food composition, and generally ranges from about 0.01% to 10% by weight.

The powder food composition contains from 35% to 78% by weight of carbohydrates. The carbohydrates may be any of the digestible carbohydrates such as dextrose, fructose, sucrose, maltose, oligosaccharides, high saccharides, or mixtures thereof, depending on usage.

Vitamins, minerals, and other trace elements can be used to supplement the food composition and for purposes

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of total nutritional balance. These supplements can be varied as desired by are typically equal to the RDA or greater based on 2,000 calories. Soy bran, rice bran, or other fiber polysaccharides or sources of fiber can be included in the food composition.

The powdered food composition includes 0.1% to 20%, preferably 0.2% to 5.0%, by weight of a complex carbohydrate stabilizer selected from the group consisting of complex carbohydrates and carbohydrate derivatives which function to prevent the precipitation of protein when a food drink composition formulated in accordance with the invention is sterilized to kill all microorganisms in the food drink composition. The complex carbohydrates and carbohydrate derivatives are typically of plant origin and function as a stabilizer which prevents the coagulation, clustering, and precipitation of protein in high temperature acidic conditions. Complex colloidal carbohydrate derivatives are presently preferred and comprise pectic substances containing a large proportion of units (in excess of 50% by weight of the pectic substance) derived from galacturonic acid and subdivided into protopectins, pectins, pectinic acids, and pectic acids. The presently preferred pectic substance is pectin.

Conventional coloring agents, such as the FDA colors, may be used, as well as conventional preservatives, such as BHT and BHA. BHT and BHA preserve fats.

The food composition is provided in a powdered form having a relatively low moisture content. The moisture content is, as is the case for many powdered formulations, preferably at least below 4% by weight and more preferably below 3% by weight. Such low moisture content provides a product having a shelf life of at least one year shelf stability at ambient conditions if hermetically sealed.

The powdered form of the food composition may be reconstituted with a liquid. The liquid form of the food composition of the invention need not be pasteurized or stored under refrigerated conditions. However, in one preferred form of the invention, the liquid form is

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sterilized at a temperature of a least 200 degrees Fahrenheit. During this sterilization process, a novel low pH protein stabilizer system comprised of a pectic substance and methylcellulose prevents the precipitation of protein from the liquid at high temperatures. This stabilizer system is described in detail later herein.

The dried powder is reconstituted with any desired edible liquid. The powder is ordinarily partially dissolved and partially suspended in the resulting liquid form of the invention. While it is possible to reconstitute the composition with liquid such as alcohol, the reconstituting liquid will ordinarily be principally water. The water may contain additional ingredients such as alcohol, glycerol, propylene glycol, sugars and flavor.

The caloric content of the liquid solutions of the reconstituted food composition of the invention is adjusted to any desired level up to about 3 calories per cubic centimeter. One half to two calories per cubic centimeter is preferred.

The osmolarity of the reconstituted food composition is in the range of 250 to 650, but preferably is in the range of 275 to 350 mOSm per liter of 1 calorie per cubic centimeter food.

The powder food compositions also include 0.1% to 8% by weight edible acidulants such as malic acid, acetic acid, citric acid, lactic acid, sodium acetate, fumaric acid, or an acidic salt such as sodium acetate in order to adjust the pH within the range of 2 to 6.5, preferably about 3 to 5.7. This pH is critical to the extended shelf life of the invention. Any pH in excess of about 6.5 is not preferred because such allows greater microbial activity and minimizes the antimicrobial effects of sorbates and benzoates utilized in the invention. A pH greater than 6.5 is totally unacceptable because of the greatly reduced antimicrobial activity of the sorbates and benzoates critical to the invention.

The antimicrobial activity of sorbic and benzoic acid is due primarily to the undissociated acid molecule.

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Antimicrobial activity is therefore pH dependent and the estimated activity at any pH can be estimated as shown below in Table 1.

TABLE 1
EFFECT OF pH ON DISSOCIATION

	pH	Percent Undissociated Acid	
		Sorbic	Benzoic
	3	98	94
	4	86	60
5	5	37	13
	6	6	1.5
10	7	0.6	0.15

The food composition includes 0.01% to 6% by weight of a sorbate or benzoate such as sorbic acid, benzoic acid, potassium sorbate, sodium sorbate, potassium benzoate, sodium benzoate, and the like. Such benzoates and sorbates are crucial because at low pH values in the range of 2 to 6.5 they provide significant antimicrobial activity.

A novel low pH protein stabilizer system is utilized in the food composition of the invention. Various conventional protein stabilizer systems will not function properly at the high temperatures used to sterilize the reconstituted food composition and permit the protein to precipitate out of the reconstituted composition. I have discovered a novel low pH protein stabilizer system which effectively stabilizes the protein in the reconstituted food composition of the invention at high temperature. The low pH protein stabilizer food composition of the invention also appears to produce an interactive synergistic effect which causes the bacteria count in the reconstituted food composition to be low when the food composition is permitted to set exposed to the air at room temperature.

The low pH protein stabilizer system in the powder form of the invention includes, as described above, from 4% to 22% on a dry weight basis whey protein or other protein alpha-amino acid and from 0.1% to 20.0%, preferably 0.2% to 5.0%, by dry weight of pectin or another pectic substance.

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In addition, .001% to 10.0%, preferably 0.01% to 4.0%, by weight of sodium carboxymethylcellulose or another methylcellulose is preferably, but not necessarily, utilized with the pectic substance because the carboxymethylcellulose and pectic substance synergistically interact to effectively stabilize the aqueous food composition which results when the powder form is reconstituted with water. Utilizing whey protein in the food composition without pectin, with or without carboxymethylcellulose, is not acceptable because the food composition, when reconstituted, does not exhibit the ability to prevent the whey protein from precipitating out of the reconstituted solution at high sterilization temperatures. When, however, whey protein is utilized in combination with pectin and sodium carboxymethylcellulose a reconstituted food composition results which is unusually stable at high temperature and resistant to the growth of aerobic and anaerobic bacteria. Samples of the reconstituted food composition of the invention have been left exposed to the air for ten days and with the detection of fewer than 10 to 20 aerobic bacteria per millimeter of reconstituted food composition. The growth of so few bacteria is highly unusual. Further, when bacteria were "dumped" into the reconstituted food composition, the number of such bacteria gradually decreased over time until living bacteria no longer existed.

After the dried powder food composition of the invention is reconstituted it has an extended shelf life at room temperature of several days or more. The ratio of water to composition will vary with the proportion of the ingredients of the composition and with the desired consistency required, as discussed above. By way of example, on a weight/weight basis of composition to water, the dilutions on a 100 gram basis can be:

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	Calories/ml. of solution	To make 100 grams solution gms powder*/gms water	Approximate Viscosity (centipoises)
	0.5	18/82	<100
5	1	25/75	100
	1.5	32/68	250
	2	40/50	500

*Powder of Example 1 below

The following examples depict the presently preferred embodiments of the invention for the purposes of illustrating the practice thereof and not by way of limitation of the scope of the invention. In the examples, all proportions are by weight, unless otherwise noted.

EXAMPLE 1

The food composition in powder form was prepared by blending a number of ingredients.

	<u>Component</u>	<u>Pounds</u>
	SUGAR	1592.000
	WHEY PROTEIN CONC., (FORETEIN 35) (1)	660.700
20	CALCIUM LACTATE, PENTAHYDRATE (4)	76.100
	NON DAIRY CREAMER (CREATIVE CREAMER 829) (2)	54.300
	MALTODEXTRIN, M100 (POLYSACCHARIDES)	51.000
	CITRIC ACID	43.500
	SODIUM CARBOXYMETHYLCELLULOSE (3)	10.000
25	PECTIN-CITRIC	35.000
	EMULSIFIER (BEATREME 3581Z) (2)	5.200
	SODIUM CITRATE	4.300
	BETA CAROTENE, 1% DILUTION CWS	4.300
	BIOTIN	0.005
30	CALCIUM PANTOTHENATE	0.180
	FERRIC ORTHOPHOSPHATE, DIHYDRATE	0.900
	FOLIC ACID	0.007
	MANGANESE SULFATE, MONOHYDRATE	0.100
	NIACINAMIDE	0.316
35	POTASSIUM SORBATE	4.544
	SELENIUM YEAST CONCENTRATE	0.900
	VITAMIN B-1 MONONITRATE	0.025
	VITAMIN B-12 1% DILUTION	0.010
	VITAMIN B-2 TYPE S	0.028

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	VITAMIN B-6 HCL	0.040
	VITAMIN C	3.190
	VITAMIN D3, 100 S.D.	0.100
	VITAMIN E 50% S.D.	0.990
5	ZINC SULFATE, MONOHYDRATE	0.700
	FLAVOR (Q.S.)	13.000
	TOTAL	2561.435

10 (1) Wisconsin Dairies, Foremost Ingredients Group, Box 111, Baraboo, Wisconsin 53913-0111; (608) 356-8316.

(2) Beatreme Foods, 352 East Grand Avenue, Beloit, Wisconsin 53511; (800) 328-7517.

15 (3) Aqualon Company, Little Falls Centre One, 2711 Centerville Road, Wilmington, Delaware 19850; (800) 345-8104.

(4) Gallard-Schlesinger Industries, 584 Mineola Avenue, Carle Place, New York 11514; (516) 333-5600.

20 The approximate percent calories from the various ingredients are carbohydrates 50.0%, fat 10.0%, and protein 40.0%. The carbohydrates included in the powder food composition include sucrose, dextrose, maltose, lactose, trisaccharides, tetrasaccharides, pentasaccharides, hexasaccharides, and higher saccharides. When 25 gm of the food powder composition is reconstituted with 75 gm of water
25 the resulting mixture has a caloric density (Cal/ml) of about 1.0; and, a total Cal/Nitrogen ratio of about 160.

30 During the blending of the above-listed ingredients of the food composition, agglomeration techniques are preferably employed to make the resulting powder mixture more easily dispersed and soluble in water.

EXAMPLE 2

35 The 2561.435 pounds of the food composition powder of Example 1 is mixed with 6538.000 pounds of water. The resulting drink provides 1.1 calories per cubic centimeter, has a pH of about 4.7, has an osmolarity of 300, has a viscosity of about 90 to 100 centipoise, and has particles each having a size of less than about 100 mesh.

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EXAMPLE 3

One thousand grams of a food composition in powder form is prepared by blending the following ingredients in the proportions noted.

5	INGREDIENT	WEIGHT PERCENT
		Dry
	SUGAR	5.5
	WHEY PROTEIN CONCENTRATE	13.35
	FORETEIN 35 (protein alpha-amino acids)	
10	CALCIUM LACTATE, PENTAHYDRATE	3.67
	CREATIVE CREAMER 829 (fat emulsifier)	5.5
	MALTODEXTRIN, M100 (agglomerated)	58.03
	CITRIC ACID	2.2
	EMULSIFIER, BEATREME 3581Z (fat emulsifier)	.22
15	SODIUM CITRATE	.21
	VITAMIN PREMIX	.22
	(Vitamins A, D, C, K, etc.)	
	MAGNESIUM OXIDE	.18
	POTASSIUM SORBATE	.46
20	PECTIN	10.00
	WATER	.46
		100.00

The approximate percent calories from the various ingredients are carbohydrates 50%, fat 10%, and protein 40%. The carbohydrates included in the powder food composition include sucrose, dextrose, maltose, lactose, trisaccharides, tetrasaccharides, pentasaccharides, hexasaccharides, and higher saccharides. When 25 gm of the food powder composition is reconstituted with 75 gm of water the resulting mixture has a caloric density (Cal/ml) of about 1.

EXAMPLE 4

Two hundred and thirty-seven grams of food composition powder of EXAMPLE 3 is mixed at 76 degrees F with 832 milliliters of sterile distilled water at 5:00 pm on May 11, 1992. The resulting drink provides about 1 calorie per cubic centimeter, has a pH of about 4.6, has an osmolarity of about 300, has a viscosity of about 90 to 100 centipoise, has a total acidity of about 5.85, and has

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particulate each having a size of less than about 100 mesh. Total acidity equals the milliliters of a one MOL NaOH aqueous solution required to neutralize the acid pH of one liter of the resulting drink. The acid pH of the drink is
5 neutralized when a pH of 7 is obtained while titrating the drink with a one MOL NaOH aqueous solution. The total acidity of the drink of the invention is increased by utilizing a buffer system which resists any change in the pH of the drink. In particular, the sodium citrate and citric
10 acid found in the compositions of Examples 1 and 3 increase the total acidity of the compositions. While any desired buffer system can be utilized in the practice of the invention, sodium citrate, citric acid, and/or sodium salts are, by way of example and not limitation, presently
15 preferred. In the practice of the invention the total acidity of the resulting drink is greater than four, preferably is greater than five, and under the most preferred conditions is greater than 5.4. The pH of the resulting drink is equal to or less than about 4.9, is
20 preferably equal to or less than about 4.75, and under the most preferred conditions is less than about 4.5. The combination of a low pH and high total acidity has been found important in providing a drink which has a high antimicrobial capacity. Other comparable prior art
25 compositions do not utilize the pH--total acidity combination of the invention. For example the PRECISION LR DIET (Trademark) composition marketed by Sandoz has a pH of about 6.8 and a total acidity of about 3.6; the TOLEREX (Trademark) composition marketed by Norwich-Eaton has a pH
30 of about 5.5 and a total acidity of about 3.2; the VITAL HIGH NITROGEN (Trademark) composition marketed by Ross Laboratories has a pH of about 6.7 and a total acidity of about 3.2; and, the VIVONEX (Trademark) composition marketed by Norwich Eaton has a pH of about 5.3 and a total acidity
35 of about 5.1.

EXAMPLE 5

One thousand grams of the powder of EXAMPLE 3 is prepared by mixing the ingredients in the proportions noted,

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except 0.23 grams of potassium sorbate is substituted for the 0.46 grams of potassium sorbate.

EXAMPLE 6

Two hundred and thirty seven grams of the food composition powder of EXAMPLE 5 is mixed at 76 degrees F with 832 milliliters of sterile distilled water at 5:00 pm on May 11, 1992. The resulting drink provides about 1 calorie per cubic centimeter, has a pH of about 4.6, has an osmolarity of about 300, has a viscosity of about 90 to 100 centipoises, has a total acidity of about 5.85, and has particles of food composition each having a size of less than about 100 mesh.

EXAMPLE 7

As soon as the drink (suspension) of EXAMPLE 4 is produced, i.e., as soon as the rehydration of the powder is performed, a plate count is performed to determine the presence of aerobic and anaerobic bacteria. The plate count is performed by transferring one milliliter of the drink to a 10 milliliter enriched Thio. The Thio is incubated at 35°C for four days to culture for anaerobes. The Thio is then examined to determine the existence of aerobic and anaerobic bacteria. The forgoing plate count procedure is carried out in accordance with the FDA Bacteriological Analytical Manual, 4th Edition, 1984, Chapter 4, and with the ASM Manual of Clinical Microbiology, 4th Edition, 1985.

The drink of EXAMPLE 4 is stored at room temperature exposed to the air. A plate count is initiated at 5:00 pm each day for ten consecutive days. As shown below in TABLE II, in each plate count less than ten aerobic microorganisms (bacteria) per grams are detected. No anaerobic bacteria are detected during any of the plate counts.

TABLE II

PLATE COUNT RESULTS SHOWING ABSENCE OF
AEROBIC BACTERIA IN SUSPENSION

PLATE COUNT FOR DAY NO.	DESCRIPTION	AEROBIC ORGANISMS PER MILLILITER
1	Rehydration, 5:00pm	<10

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	2	1 day, 5:00pm	<10
	3	2 days, 5:00pm	<10
	4	3 days, 5:00pm	<10
	5	4 days, 5:00pm	<10
5	6	5 days, 5:00pm	<10
	7	6 days, 5:00pm	<10
	8	7 days, 5:00pm	<10
	9	8 days, 5:00pm	<10
	10	9 days, 5:00pm	<10
10	11	10 days, 5:00pm	<10

EXAMPLE 8

As soon as the drink (suspension) of EXAMPLE 6 is produced, i.e., as soon as the rehydration of the powder is performed, a plate count is performed to determine the presence of aerobic and anaerobic bacteria. The plate count is performed by transferring one milliliter of the drink to a 10 milliliter enriched Thio. The Thio is incubated at 35°C for four days to culture for anaerobes. The Thio is then examined to determine the existence of aerobic and anaerobic bacteria. The foregoing plate count procedure is carried out in accordance with the FDA Bacteriological Analytical Manual, 4th Edition, 1984, Chapter 4, and with the ASM Manual of Clinical Microbiology, 4th Edition, 1985.

The drink of EXAMPLE 6 is stored at room temperature. A plate count is initiated at 5:00pm each day for ten consecutive days. As shown below in TABLE III, in each plate count twenty or less aerobic organisms per gram are detected.

TABLE III

PLATE COUNT RESULTS SHOWING ABSENCE OF
AEROBIC BACTERIA IN SUSPENSION

	PLATE COUNT FOR DAY NO.	DESCRIPTION	AEROBIC ORGANISMS PER MILLILITER
35	1	Rehydration, 5:00pm	20
	2	1 day, 5:00pm	<10
	3	2 days, 5:00pm	<10
	4	3 days, 5:00pm	<10

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	5	4 days, 5:00pm	<10
	6	5 days, 5:00pm	<10
	7	6 days, 5:00pm	<10
	8	7 days, 5:00pm	<10
5	9	8 days, 5:00pm	<10
	10	9 days, 5:00pm	<10
	11	10 days, 5:00pm	<10

EXAMPLE 9

10 Two hundred and thirty seven grams of the food
composition powder of EXAMPLE 3 was mixed with 832
milliliters of sterile distilled water in a beaker to form
a homogenous solution. The solution had a viscosity of about
250 centipoises, The size of food composition particles in
15 suspension in the solution was less than 100 mesh. The
solution was allowed to stand for ten days at room
temperature. At the end of the ten day period, the solution
was still substantially homogeneous and particulate had not
settled or separated out of solution form layers of material
20 at the bottom of the beaker.

EXAMPLE 10

 Two hundred and thirty seven grams of the food
composition powder of EXAMPLE 5 was mixed with 832
milliliters of sterile distilled water in a beaker to form
25 a homogeneous solution. The solution has a viscosity of
about 250 centipoises. The size of the food composition
particles in suspension in the solution was less than or
equal to 100 mesh. The solution was allowed to stand for ten
days at room temperature. At the end of the ten day period,
30 the solution was still substantially homogeneous and
particulate had not settled or separated out of solution to
form layers of material at the bottom of the beaker.

EXAMPLE 11

 One thousand grams of the powder of EXAMPLE 3 is
35 prepared by mixing the ingredients in the proportions noted,
except that 8.0 grams of Maltodextrin, M100 and 2.0 grams of
whey protein concentrate foretein 35 are utilized in place
of the ten grams of pectin.

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EXAMPLE 12

Two hundred and thirty seven grams of the food composition powder of EXAMPLE 11 is mixed with 832 milliliters of sterile distilled water in a beaker to form a homogeneous solution. The solution has a viscosity of about 250 centipoises. The size of the food composition particles in suspension in the solution is less than or equal to 100 mesh. The solution is allowed to stand at room temperature. In less than six hours particulate begin settling and separating out of the solution to form a layer of material at the bottom of the beaker.

EXAMPLE 13

One thousand grams of the powder of EXAMPLE 3 is prepared by mixing the ingredients in the proportions noted, except that 8.0 grams of Maltodextrin, M100 and 2.0 grams of whey protein concentrate FORETEIN 35 are utilized in place of the ten grams of pectin.

EXAMPLE 14

Two hundred and thirty seven grams of food composition powder of EXAMPLE 13 is mixed with 832 milliliters of sterile distilled water in a beaker to form a homogeneous solution. The solution has a viscosity of about 250 centipoises. The size of food composition particles in suspension in the solution is less than or equal to 100 mesh. The solution is heated to and remains at the temperature of 220 degrees Fahrenheit for five (5) minutes. After the solution had been heated for two minutes, precipitate begins to form and fall to the bottom of the beaker. The precipitate contains whey protein.

EXAMPLE 15

Two hundred and thirty seven grams of food composition powder of EXAMPLE 3 is mixed with 832 milliliters of sterile distilled water in a beaker to form a homogeneous solution. The solution has a viscosity of about 250 centipoises. The size of food composition particles in suspension in the solution is less than or equal to 100 mesh. The solution is heated to and remains at the temperature of 220 degrees Fahrenheit for five (5)

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minutes. Precipitate does not form during the time the solution is heated.

EXAMPLE 16

Two hundred and thirty seven grams of food composition powder of EXAMPLE 1 is mixed with 832 milliliters of sterile distilled water in a beaker to form a homogeneous solution. The solution has a viscosity of about 250 centipoises. The size of food composition particles in suspension in the solution is less than or equal to 100 mesh. The solution is heated to the temperature of 285 degrees Fahrenheit for ten (10) seconds. Precipitate does not form during the time the solution is heated.

EXAMPLE 17

Two hundred and thirty seven grams of food composition powder of EXAMPLE 13 is mixed with 832 milliliters of sterile distilled water in a beaker to form a homogeneous solution. The solution has a viscosity of about 250 centipoises. The size of food composition particles in suspension in the solution is less than or equal to 100 mesh. The solution is heated to and remains at the temperature of 300 degrees Fahrenheit for four (4) seconds. After the solution had been heated for one (1) minutes, precipitate begins to form and fall to the bottom of the beaker. The precipitate contains whey protein.

EXAMPLE 18

One thousand grams of the powder of EXAMPLE 1 is prepared by mixing the ingredients in the proportions noted, except that 1.0 gram of Maltodextrin (Polysaccharides) is utilized in place of the one gram of pectin.

EXAMPLE 19

Two hundred and thirty seven grams of food composition powder of EXAMPLE 18 is mixed with 832 milliliters of sterile distilled water in a beaker to form a homogeneous solution. The solution has a viscosity of about 250 centipoises. The size of food composition particles in suspension in the solution is less than or equal to 100 mesh. The solution is heated to and remains at the temperature of 300 degrees Fahrenheit for four (4)

- 20 -

seconds. After the solution had been heated for one (1) minute, precipitate begins to form and fall to the bottom of the beaker. The precipitate contains whey protein.

EXAMPLE 20

5 Two hundred and thirty seven grams of food composition powder of EXAMPLE 18 is mixed with 832 milliliters of sterile distilled water in a beaker to form a homogeneous solution. The solution has a viscosity of about 250 centipoises. The size of food composition
10 particles in suspension in the solution is less than or equal to 100 mesh. The solution is heated to and remains at the temperature of 285 degrees Fahrenheit for ten (10) seconds. After the solution has been heated for two (2) minutes, precipitate begins to form and fall to the bottom
15 of the beaker. The precipitate contains whey protein.

EXAMPLE 21

 Two hundred and thirty seven grams of food composition powder of EXAMPLE 1 is mixed with 832 milliliters of sterile distilled water in a beaker to form
20 a homogeneous solution. The solution has a viscosity of about 250 centipoises. The size of food composition particles in suspension in the solution is less than or equal to 100 mesh. The solution is heated to and remains at the temperature of 300 degrees Fahrenheit for four (4)
25 seconds. Precipitate does not form during the time the solution is heated.

EXAMPLE 22

 One thousand grams of the powder of EXAMPLE 3 is prepared by mixing the ingredients in the proportions noted,
30 except that ten grams of protopectin is utilized in place of the ten grams of pectin.

EXAMPLE 23

 Two hundred and thirty seven grams of food composition powder of EXAMPLE 22 is mixed with 832
35 milliliters of sterile distilled water in a beaker to form a homogeneous solution. The solution has a viscosity of about 250 centipoises. The size of food composition particles in suspension in the solution is less than or

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equal to 100 mesh. The solution is heated to and remains at the temperature of 285 degrees Fahrenheit for ten (10) seconds. Precipitate does not form during the time the solution is heated.

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EXAMPLE 24

One thousand grams of the powder of EXAMPLE 3 is prepared by mixing the ingredients in the proportions noted, except that ten grams of pectinic acid is utilized in place of the ten grams of pectin.

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EXAMPLE 25

Two hundred and thirty seven grams of food composition powder of EXAMPLE 24 is mixed with 832 milliliters of sterile distilled water in a beaker to form a homogeneous solution. The solution has a viscosity of about 250 centipoises. The size of food composition particles in suspension in the solution is less than or equal to 100 mesh. The solution is heated to the temperature of 300 degrees Fahrenheit for three (3) seconds. Precipitate does not form during the time the solution is heated.

20

EXAMPLE 26

Examples 24 and 25 are repeated in sequence, except that in Example 24 the amount of pectinic acid in the powder is reduced to a weight percent sufficient to permit some precipitate containing whey protein to form during Example 25.

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EXAMPLE 27

Example 26 is repeated, except the powder of Example 24 includes 0.40% by weight sodium carboxymethylcellulose. No precipitate is formed during Example 25.

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EXAMPLE 28

Example 27 is repeated, except in Example 24 ten grams of pectin is substituted for ten grams of pectinic acid. Similar results are obtained.

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EXAMPLE 29

Examples 3 and 4 are repeated, except that the amount of potassium sorbate in the food composition prepared

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in Example 3 is reduced such that the pH of the resulting drink in Example 4 is about 5.1 instead of about 4.6. The resulting drink still provides one calorie per cubic centimeter, has an osmolarity of about 300, has a viscosity of about 90 to 100 centipoise, and has particles each having a size of less than about 100 mesh. The total acidity of the resulting drink is about 5.05.

EXAMPLE 30

As soon as the drink (suspension) of EXAMPLE 29 is produced, i.e., as soon as the rehydration of the powder is performed, a plate count is performed to determine the presence of aerobic and anaerobic bacteria. The plate count is performed by transferring one milliliter of the drink to a 10 milliliter enriched Thio. The Thio is incubated at 35°C for four days to culture for anaerobes. The Thio is then examined to determine the existence of aerobic and anaerobic bacteria. The forgoing plate count procedure is carried out in accordance with the FDA Bacteriological Analytical Manual, 4th Edition, 1984, Chapter 4, and with the ASM Manual of Clinical Microbiology, 4th Edition, 1985.

The drink of EXAMPLE 29 is stored at room temperature exposed to the air. A plate count is initiated at 5:00 pm each day for ten consecutive days following the day on which the drink of EXAMPLE 29 is formulated by rehydrating the powder food composition powder. In the plate count taken at rehydration and in each of the ten plate counts taken after rehydration, in excess of ten aerobic microorganisms per grams are detected. Anaerobic bacteria are detected during each of the plate counts.

EXAMPLE 31

Examples 3 and 4 are repeated, except that the amount of potassium sorbate in the food composition prepared in Example 3 is reduced such that the pH of the resulting drink in Example 4 is about 4.9 instead of about 4.6. The resulting drink still provides 1 calorie per cubic centimeter, has an osmolarity of about 300, has a viscosity of about 90 to 100 centipoise, and has particles each having

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a size of less than about 100 mesh. The total acidity of the resulting drink is about 5.4.

EXAMPLE 32

As soon as the drink (suspension) of EXAMPLE 31 is produced, i.e., as soon as the rehydration of the powder is performed, a plate count is performed to determine the presence of aerobic and anaerobic bacteria. The plate count is performed by transferring one milliliter of the drink to a 10 milliliter enriched Thio. The Thio is incubated at 35°C for four days to culture for anaerobes. The Thio is then examined to determine the existence of aerobic and anaerobic bacteria. The forgoing plate count procedure is carried out in accordance with the FDA Bacteriological Analytical Manual, 4th Edition, 1984, Chapter 4, and with the ASM Manual of Clinical Microbiology, 4th Edition, 1985.

The drink of EXAMPLE 31 is stored at room temperature exposed to the air. A plate count is initiated at 5:00 pm each day for ten consecutive days following the day on which the drink of EXAMPLE 31 is formulated by rehydrating the powder food composition powder. In the plate count taken at rehydration and in each of the ten plate counts taken after rehydration, less than ten aerobic organisms per gram are detected. Anaerobic bacteria are not detected during any of the plate counts.

EXAMPLE 33

Examples 3 and 4 are repeated, except that the amount of potassium sorbate in the food composition prepared in Example 3 is increased such that the pH of the resulting drink in Example 4 is about 4.0 instead of about 4.6. The resulting drink still provides 1 calorie per cubic centimeter, has an osmolarity of about 300, has a viscosity of about 90 to 100 centipoise, and has particles each having a size of less than about 100 mesh. The resulting drink has a total acidity of about 5.9.

EXAMPLE 34

As soon as the drink (suspension) of EXAMPLE 33 is produced, i.e., as soon as the rehydration of the powder is performed, a plate count is performed to determine the

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presence of aerobic and anaerobic bacteria. The plate count is performed by transferring one milliliter of the drink to a 10 milliliter enriched Thio. The Thio is incubated at 35°C for four days to culture for anaerobes. The Thio is then
5 examined to determine the existence of aerobic and anaerobic bacteria. The forgoing plate count procedure is carried out in accordance with the FDA Bacteriological Analytical Manual, 4th Edition, 1984, Chapter 4, and with the ASM Manual of Clinical Microbiology, 4th Edition, 1985.

10 The drink of EXAMPLE 33 is stored at room temperature exposed to the air. A plate count is initiated at 5:00 pm each day for ten consecutive days following the day on which the drink of EXAMPLE 33 is formulated by rehydrating the powder food composition powder. In the
15 plate count taken at rehydration and in each of the ten plate counts taken after rehydration, less than ten aerobic microorganisms (bacteria) per gram are detected. Anaerobic bacteria are not detected during any of the plate counts.

EXAMPLE 35

20 Examples 3 and 4 are repeated, except that the amount of potassium sorbate in the food composition prepared in Example 3 is increased such that the pH of the resulting drink in Example 4 is about 3.0 instead of about 4.6. The
25 resulting drink still provides 1 calorie per cubic centimeter, has an osmolarity of about 300, has a viscosity of about 90 to 100 centipoise, and has particles each having a size of less than about 100 mesh. The resulting drink has a total acidity of about 6.10.

EXAMPLE 36

30 As soon as the drink (suspension) of EXAMPLE 35 is produced, i.e., as soon as the rehydration of the powder is performed, a plate count is performed to determine the presence of aerobic and anaerobic bacteria. The plate count is performed by transferring one milliliter of the drink to
35 a 10 milliliter enriched Thio. The Thio is incubated at 35°C for four days to culture for anaerobes. The Thio is then examined to determine the existence of aerobic and anaerobic bacteria. The forgoing plate count procedure is carried out

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in accordance with the FDA Bacteriological Analytical Manual, 4th Edition, 1984, Chapter 4, and with the ASM Manual of Clinical Microbiology, 4th Edition, 1985.

5 The drink of EXAMPLE 35 is stored at room temperature exposed to the air. A plate count is initiated at 5:00 pm each day for ten consecutive days following the day on which the drink of EXAMPLE 35 is formulated by rehydrating the powder food composition powder. In the plate count taken at rehydration and in each of the ten
10 plate counts taken after rehydration, less than ten aerobic organisms per gram are detected. Anaerobic bacteria are not detected during any of the plate counts.

While preparing drinks in accordance with the invention which have a pH in the range of 2 to 6.5 helps
15 prevent or minimize the formation of aerobic and anaerobic bacteria in the drinks, I have discovered that producing a drink with a pH in the range of 2 to about 4.9 is critical in preventing the growth of all or substantially all bacteria in the drink. As used herein, the growth of
20 "substantially all" microorganisms (bacteria) is prevented if a concentration of no more than ten aerobic organisms per milliliter results in the drink at room temperature.

EXAMPLE 37

Examples 31 and 32 are repeated, except that the
25 quantity of sodium citrate and citric acid in the drink is reduced to lower the total acidity from 5.4 to about 5.00. Similar results are obtained.

EXAMPLE 38

Examples 33 and 34 are repeated, except that the
30 quantity of sodium citrate and citric acid in the drink is reduced to lower the total acidity from 5.9 to about 5.00. Similar results are obtained.

EXAMPLE 39

Examples 35 and 36 are repeated, except that the
35 quantity of sodium citrate and citric acid in the drink is reduced to lower the total acidity from 6.1 to about 4.5. Similar results are obtained.

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The viscosity of the food composition of the invention is important and is, in part, responsible for the difficulty in finding a suitable stabilizer. The viscosity is such that the food composition, when reconstituted with water, can be readily drunk. The viscosity at 68°F of the reconstituted food composition is less than 10,000 centipoises, preferably less than 1000 centipoises. The viscosity of olive oil at 68°F is 1008 millipoises; of sperm oil at 68°F is 420 millipoises; of water at 68°F is 10.02 millipoises; of castor oil at 68°F is 10,272 millipoises; of turpentine at 68°F is 14.87 millipoises; of methyl alcohol at 68°F is 5.93 millipoises; and, of glycerol at 20°C is 10,690 millipoises. The viscosity of glycerol at 20.9°C is 7,776 millipoises. Even at low viscosities of 500 centipoises or less, the food composition of the invention retains its homogeneity. In one embodiment of the invention, the preferred viscosity is less than 500 centipoises.

The size of the particles in the food composition of the invention after the food composition is reconstituted is also important. Particles in the reconstituted food composition generally are each equal to or less than 100 mesh in size. A 20 mesh particle moves through a screen opening of 0.0331 inch; a 50 mesh particle moves through a screen opening of 0.0117 inch; a 100 mesh particle moves through a screen opening of 0.0059 inch; a 200 mesh particle moves through a screen opening of 0.0021 inch; and, a 325 mesh particle moves through a screen opening of 0.0017 inch. Since particulate in the reconstituted food composition must remain in suspension, the particulate size is small.

The food composition of the invention is ingested at any desired point along the digestive tract, but ordinarily is administered to a patient orally or is tubally fed directly into the patient's stomach. If appropriate, the reconstituted food composition can be tubally directly fed into the intestinal tract or the esophagus. The patient can, as would be appreciated by those of skill in the art, can be a hominid or other appropriate animal.

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Having described my invention in such terms as to enable those skilled in the art to understand and practice it, and having identified the presently preferred embodiments thereof, I Claim:

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CLAIMS

1. A food composition for ingestion along the digestive tract of a patient, said food composition consisting of:
 - 5 (a) from 6% to 28% by weight of natural water soluble protein;
 - (b) from 4% to 22% by weight of triglycerides of predominantly 6 to 26 carbon atoms in the fatty acid chain;
 - 10 (c) from 35% to 78% by weight of carbohydrates selected from the group consisting of corn syrup solids, trisaccharides, tetrasaccharides, pentasaccharides, hexasaccharides, dextrose, fructose, sucrose, maltose, oligosaccharides and higher saccharides;
 - 15 (d) from 0.01% to 10% by weight of an emulsifier;
 - (e) from 0.1% to 8% by weight of an edible acid for adjusting the pH of the food composition within the range of 2 to about 4.9 when said food composition is hydrated;
 - 20 (f) from 0.01% to 6% by weight of an antimicrobial agent selected from the group consisting of sorbic acid, benzoic acid, sodium benzoate, potassium sorbate, sodium sorbate, and potassium benzoate; and,
 - 25 (g) from 1% to 5.0% by weight of a water soluble complex carbohydrate stabilizer selected from the group consisting of complex carbohydrates and complex carbohydrate derivatives, said stabilizer preventing the precipitation of said protein when said food composition is hydrated to form a drink and is sterilized to kill substantially all microorganisms in the drink.
2. The food composition of Claim 1 in liquid form and
 - 35 (a) including water;
 - (b) providing from 0.4 up to about 3 calories per cubic centimeter of composition;
 - (c) having an osmolarity from 250 up to about 650;

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- (d) having a viscosity of less than 500 centipoise;
and,
(e) including particulate, substantially all of said
particulate being less than about 100 mesh in
size.
- 5 3. The food composition of Claim 1 wherein said water
soluble complex carbohydrate stabilizer is pectin.
4. The food composition of Claim 2 wherein said water
soluble complex carbohydrate stabilizer is pectin.
- 10 5. A method for preparing a sterile food composition for
ingestion along the digestive tract of a patient, said
method including the steps of
- (a) preparing a powder food composition by blending
together
- 15 (i) from 6% to 28% by weight of water soluble
protein;
- (ii) from 4% to 22% by weight of triglycerides of
predominantly 6 to 26 carbon atoms in the
fatty acid chain;
- 20 (iii) from 35% to 78% by weight of carbohydrates
selected from the group consisting of corn
syrup solids, trisaccharides,
tetrasaccharides, pentasaccharides,
hexasaccharides, dextrose, fructose, sucrose,
- 25 maltose, oligosaccharides and higher
saccharides;
- (iv) from 0.1% to 10% by weight of an emulsifier;
- (v) from 0.1% to 8% by weight of an edible acid
for adjusting the pH of the food composition
within the range of 2 to about 4.9 when said
food composition is hydrated;
- 30 (vi) from 0.01% to 6% by weight of an
antimicrobial agent selected from the group
consisting of sorbic acid, benzoic acid,
sodium benzoate, potassium sorbate, sodium
sorbate, and potassium benzoate; and,
- 35 (v) from 1% to 5.0% by weight of a water soluble
complex carbohydrate stabilizer selected from

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- 5 the group consisting of complex carbohydrates and complex carbohydrate derivatives, said stabilizer preventing the precipitation of said protein when said food composition is hydrated to form a drink and is sterilized to kill substantially all microorganisms in the drink;
- (b) mixing said powder food composition with water to form an aqueous food solution;
- 10 (c) heating said aqueous food solution to a sterilization temperature of at least about two hundred degrees Fahrenheit for a time sufficient to kill substantially all microorganisms in said food solution; and,
- 15 (d) cooling said sterilized food solution.
6. The method of Claim 5 wherein in step (a), from 0.001% to 4.0% by weight methylcellulose is blended into said powder food composition.
7. The method of Claim 5 wherein in step (a), from 0.001% to 4.0% by weight sodium carboxymethylcellulose is blended into said powder food composition.
- 20 8. The food composition of Claim 1 including from 0.001% to 4.0% by weight methylcellulose.
9. The food composition of Claim 1 including from 0.001% to 4.0% by weight sodium carboxymethylcellulose.
- 25 10. The method of Claim 6 wherein in step (a) said water soluble complex carbohydrate stabilizer is pectin.
11. The method of Claim 7 wherein in step (a) said water soluble complex carbohydrate stabilizer is pectin.
- 30 12. The method of Claim 5 wherein in step (a) said water soluble complex carbohydrate stabilizer is pectin.
13. A food composition for ingestion along the digestive tract of a patient, said food composition consisting of:
- 35 (a) from 6% to 28% by weight of water soluble protein;
- (b) from 4% to 22% by weight of triglycerides of predominantly 6 to 26 carbon atoms in the fatty acid chain;

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- 5 (c) from 35% to 78% by weight of carbohydrates selected from the group consisting of corn syrup solids, trisaccharides, tetrasaccharides, pentasaccharides, hexasaccharides, dextrose, fructose, sucrose, maltose, oligosaccharides and higher saccharides;
- 10 (d) from 0.01% to 10% by weight of an emulsifier;
- (e) from 0.1% to 8% by weight of an edible acid for adjusting the pH of the food composition within the range of 2 to about 4.9 when said food composition is hydrated;
- 15 (f) from 0.01% to 6% by weight of an antimicrobial agent selected from the group consisting of sorbic acid, benzoic acid, sodium benzoate, potassium sorbate, sodium sorbate, and potassium benzoate; and,
- 20 (g) from 1% to 5.0% by weight of a water soluble complex carbohydrate stabilizer selected from the group consisting of complex carbohydrates and complex carbohydrate derivatives, said stabilizer preventing the precipitation of said protein when said food composition is hydrated to form a drink and is sterilized to kill substantially all microorganisms in the drink.
- 25 14. The food composition of Claim 13 wherein said protein is produced by hydrolyzing naturally occurring protein.
15. The food composition of Claim 13 in liquid form and
- 30 (a) including water;
- (b) providing from 0.4 up to about 3 calories per cubic centimeter of composition;
- (c) having an osmolarity from 250 up to about 650; and,
- 35 (d) including particulate, substantially all of said particulate being less than about 100 mesh in size.
16. A food composition for ingestion along the digestive tract of a patient, said food composition consisting of an aqueous solution including:

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- (a) protein;
 - (b) triglycerides of predominantly 6 to 26 carbon atoms in the fatty acid chain;
 - (c) carbohydrates selected from the group consisting of corn syrup solids, trisaccharides, tetrasaccharides, pentasaccharides, hexasaccharides, dextrose, fructose, sucrose, maltose, oligosaccharides and higher saccharides;
 - (d) an emulsifier;
 - (e) an edible acid for adjusting the pH of said solution within the range of 2 to about 4.9;
 - (f) an antimicrobial agent selected from the group consisting of sorbic acid, benzoic acid, sodium benzoate, potassium sorbate, sodium sorbate, and potassium benzoate; and,
 - (g) a stabilizer;
- said solution having a total activity of at least 4.0.
17. A food composition for ingestion along the digestive tract of a patient, said food composition consisting of an aqueous solution including:
- (a) protein;
 - (b) triglycerides of predominantly 6 to 26 carbon atoms in the fatty acid chain;
 - (c) carbohydrates selected from the group consisting of corn syrup solids, trisaccharides, tetrasaccharides, pentasaccharides, hexasaccharides, dextrose, fructose, sucrose, maltose, oligosaccharides and higher saccharides;
 - (d) an emulsifier;
 - (e) an edible acid for adjusting the pH of said solution within the range of 2 to about 4.9;
 - (f) an antimicrobial agent selected from the group consisting of sorbic acid, benzoic acid, sodium benzoate, potassium sorbate, sodium sorbate, and potassium benzoate; and,
 - (g) a stabilizer;
- said solution having a total activity of at least 5.0.
18. A food composition for ingestion along the digestive

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tract of a patient, said food composition consisting of an aqueous solution including:

- (a) protein;
- (b) triglycerides of predominantly 6 to 26 carbon atoms in the fatty acid chain;
- (c) carbohydrates selected from the group consisting of corn syrup solids, trisaccharides, tetrasaccharides, pentasaccharides, hexasaccharides, dextrose, fructose, sucrose, maltose, oligosaccharides and higher saccharides;
- (d) an emulsifier;
- (e) an edible acid for adjusting the pH of said solution within the range of 2 to about 4.9;
- (f) an antimicrobial agent selected from the group consisting of sorbic acid, benzoic acid, sodium benzoate, potassium sorbate, sodium sorbate, and potassium benzoate; and,
- (g) a stabilizer;

said solution having a the total activity of at least 5.4.

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US95/02026**A. CLASSIFICATION OF SUBJECT MATTER**

IPC(6) : A23J 1/00; A23L 1/035, 1/0524, 1/0534, 1/09, 1/29

US CL : 426/335, 532, 577, 602, 607, 654, 656, 658,

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 426/335, 532, 577, 602, 607, 654, 656, 658,

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

APS

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US, A, 4,931,300 [MONTE] 05 June 1990, entire document.	1-18
X	US, A, 5,156,875 [MONTE] 20 October 1992, entire document.	1-2,5,13-18
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Y		3-4,6-12
Y	US, A, 4,112,123 [ROBERTS] 05 September 1978, entire document.	1-18
Y	US, A, 4,497,800 [LARSON ET AL] 05 February 1985, entire document.	1-18
Y	US, A, 3,950,547 [LAMAR, III ET AL] 13 April 1976, columns 2-4.	1-18

☒ Further documents are listed in the continuation of Box C. ☐ See patent family annex.

* Special categories of cited documents:	*T	later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
A document defining the general state of the art which is not considered to be part of particular relevance	*X*	document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
E earlier document published on or after the international filing date	*Y*	document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
L document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	*Z*	document member of the same patent family
O document referring to an oral disclosure, use, exhibition or other means		
P document published prior to the international filing date but later than the priority date claimed		

Date of the actual completion of the international search

23 MAY 1995

Date of mailing of the international search report

05 JUN 1995

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INTERNATIONAL SEARCH REPORTInternational application No.
PCT/US95/02026**C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT**

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US, A, 5,296,245 [CLARK ET AL] 22 March 1994, col. 2, lines 8-16, col. 4, lines 50-57.	3-4,6-12